

## REFERENCES

1. Harborne, J. B., Mabry, T. J. and Mabry, H. (eds.) (1975) *The Flavonoids*. Chapman and Hall, London.
2. Nilsson, E. and Bendz, G. (1973) *Flavonoids in Bryophytes*. In: *Chemistry in Botanical Classification, Nobel Symposium*

- 25, Bendz, G. and Santesson, J., eds. Academic Press, New York.
3. Harborne, J. B. (1963) *Phytochemistry* 2, 327.
4. Hänsel, R. and Langhammer, L. (1963) *Arch. Pharmaz.* 296, 619.

*Phytochemistry*, 1977, Vol. 16, p. 1109. Pergamon Press. Printed in England.

A NEW WOGONIN DERIVATIVE FROM *GARDENIA GUM*

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**Key Word Index**—*Gardenia lucida*; *G. gummifera*; Rubiaceae; 5,7,3',5'-tetrahydroxy-8,4'-dimethoxyflavone.

Previously, sixteen flavones have been isolated from the gum [1–6]. In continuation of our work [3–6] on the flavonoids of *Gardenia* gum, another new flavone has been isolated from the  $C_6H_6$  and  $H_2O$  insoluble portion of the alcoholic extract of the gum. The structure 5,7,3',5'-tetrahydroxy-8,4'-dimethoxyflavone assigned to it has been confirmed by synthesis.

## EXPERIMENTAL

**Extraction and isolation.** The extraction procedure has been described earlier [6]. PC (Whatman 3 mm) of  $H_2O$  insoluble portion using 50% HOAc gave 3 yellow bands,  $B_1$ ,  $B_2$  and  $B_3$ . Band  $B_1$  resolved into two bands on TLC (Si gel,  $C_6H_6$ -MeOH-AcOH, 45:3:2). The upper band furnished two known flavones A and B [5]. The lower band also resolved into two compounds C and D. Compound C has earlier been identified [5]. Bands  $B_2$  and  $B_3$  yielded seven compounds [5, 6].

**Identification.** Compound D crystallized as yellow needles, mp 254–56°;  $R_f$  0.89 (BAW, 4:1:5); 0.88. (PhOH- $H_2O$ , 3:1); 0.22 (15% aq. AcOH); (Found: C, 58.8; H, 4.4.  $C_{17}H_{14}O_8$  requires C, 59.0; H, 4.1%);  $\lambda_{max}^{MeOH}$  nm (log  $\epsilon$ ): 275 (4.03), 327 (3.88);  $AlCl_3$ , 285, 310, 350;  $AlCl_3$ -HCl, 285, 305, 345; NaOAc, 285, 340, 355; NaOAc- $H_3BO_3$ , 280, 305, 330;  $\nu_{max}^{KBr}$ : 3448, 2950, 1650, 1587, 1495, 1013 and 963  $cm^{-1}$ . Comparison of its  $R_f$  values on PC and TLC with other compounds isolated from the gum indicated it to be a tetrahydroxyflavone. It gave a positive Gibb's and a negative gossypetone test. Methylation with  $CH_2N_2$  gave a partial methyl ether, mp 215–16°;  $\lambda_{max}^{MeOH}$  nm: 275, 325, identical with 5-hydroxy-7,8,3',4',5'-pentamethoxyflavone [3] (mmp, co-TLC, UV and IR). Compound D could therefore be 5,7,3',5'-tetrahydroxy-8,4'-dimethoxyflavone.

**Synthesis of 2-(3',5'-Dibenzyloxy-4'-methoxybenzoyloxy)-4-benzyloxy-3,6-dimethoxyacetophenone.** A mixture of 2-hydroxy-4-benzyloxy-3,6-dimethoxyacetophenone [7] (700 mg), 3,5-benzyloxy-4-methoxybenzoyl chloride [8] (1.75 g) and  $C_5H_5N$  (5 ml) was heated on a steam bath for 3 hr. The cooled reaction mixture was treated with ice-HCl (1:1) and then extracted with EtOAc. The organic layer was washed with  $H_2O$ , dried ( $Na_2SO_4$ ) and concd. The brown semi-solid ester was purified by passing a soln of it in EtOAc- $C_6H_6$  (1:1) through a column of neutral alumina and crystallized from EtOAc-petrol as colourless needles (900 mg), mp 132–33°; (Found: C, 72.5; H, 5.8.  $C_{39}H_{36}O_9$  requires C, 72.2; H, 5.6%);  $\nu_{max}^{KBr}$ : 1727, 1672, 1600 and 1493  $cm^{-1}$ .

**Synthesis of 2-Hydroxy-4,3',5'-tribenzyloxy-3,6,4'-trimethoxy-dibenzoylmethane.** The above ester (700 mg) in dry  $C_5H_5N$  (7 ml)

was treated with powdered KOH (1.8 g) and the mixture shaken vigorously for 2 hr with occasional warming. The reaction mixture was worked up as above. The diketone crystallized from EtOAc-petrol as yellow needles (500 mg), mp 202–4°; (Found: C, 71.9; H, 5.7.  $C_{39}H_{36}O_9$  requires C, 72.2; H, 5.6%);  $\nu_{max}^{KBr}$  2994, 1613, 1582 and 1534  $cm^{-1}$ .

**Synthesis of 7,3',5'-Tribenzyloxy-5,8,4'-trimethoxyflavone.** The above diketone (400 mg) was gently refluxed with glacial HOAc (8 ml) and fused NaOAc (1.2 g) in an oil bath for 3 hr. The resulting flavone crystallized from EtOAc as colourless shining needles (280 mg), mp 192°; (Found: C, 74.3; H, 5.8.  $C_{39}H_{34}O_8$  requires C, 74.3; H, 5.4%);  $\lambda_{max}^{MeOH}$  nm (log  $\epsilon$ ): 275 (4.2), 330 (4.0);  $\nu_{max}^{KBr}$  1639, 1595, 1493 and 1418  $cm^{-1}$ .

**Synthesis of 7,3',5'-Trihydroxy-5,8,4'-trimethoxyflavone.** The above flavone (200 mg) in EtOAc (20 ml) was stirred in an atmosphere of  $H_2$  in the presence of Pd-C (25 mg; 10%) for 3 hr. The hydroxyflavone crystallized as yellow cubes (80 mg), mp 264–66°; (Found: C, 59.8; H, 4.30.  $C_{18}H_{16}O_8$  requires C, 60.0; H, 4.4%);  $\lambda_{max}^{MeOH}$  nm (log  $\epsilon$ ): 270 (4.4), 330 (4.3); NaOAc, 280;  $\nu_{max}^{KBr}$  3125, 1639, 1577 and 1511  $cm^{-1}$ .

**Synthesis of 5,7,3',5'-Tetrahydroxy-8,4'-dimethoxyflavone.** A mixture of the hydroxyflavone (50 mg), dry  $AlCl_3$  (95 mg) and  $CH_3CN$  (5 ml) was refluxed on a steam bath for 3 hr.  $CH_3CN$  was distilled off and the  $AlCl_3$  complex was decomposed with ice-HCl (1:1). The crude flavone was purified by preparative TLC (Si gel;  $C_6H_5CH_3$ -HCOOEt-HCOOH, 5:4:1). It crystallized from EtOH as yellow needles (20 mg), mp 255–57°; (Found: C, 58.8; H, 4.4.  $C_{17}H_{14}O_8$  requires C, 59.0; H, 4.1%). It was identical (mmp, co-TLC, UV and IR) with the natural sample.

## REFERENCES

1. Stenhouse, J. and Groves, C. E. (1877) *J. Chem. Soc.* 32, 551.
2. Rama Rao, A. V., Venkataraman, K., Chakrabarti, P., Sanyal, A. K. and Bose, P. K. (1970) *Indian J. Chem.* 8, 398.
3. Krishnamurti, M., Seshadri, T. R. and Sharma, N. D. (1972) *Indian J. Chem.* 10, 23.
4. Gupta, S. R., Seshadri, T. R., Sharma, C. S. and Sharma, N. D. (1975) *Indian J. Chem.* 13, 785.
5. Chhabra, S. C., Gupta, S. R., Seshadri, T. R. and Sharma, N. D. (1976) *Indian J. Chem.* 14B, 651.
6. Chhabra, S. C., Gupta, S. R. and Sharma, N. D. (1976) *Phytochemistry* 16, 399.
7. Geissman, T. A. and Halsall, T. G. (1951) *J. Am. Chem. Soc.* 73, 1282.
8. Schopf, C. and Winterhalder, L. (1940) *Ann.* 544, 62.